



What is Nernst Equation?

The [Nernst equation](#), a cornerstone in electrochemistry, provides a quantitative relationship between the potential of an electrochemical cell and the concentrations of the species involved in the redox reaction. This equation, named after the German physical chemist Walther Nernst, is essential for understanding and predicting the behaviour of electrochemical cells, from batteries to corrosion processes. Enroll now at Tutoroot.

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